

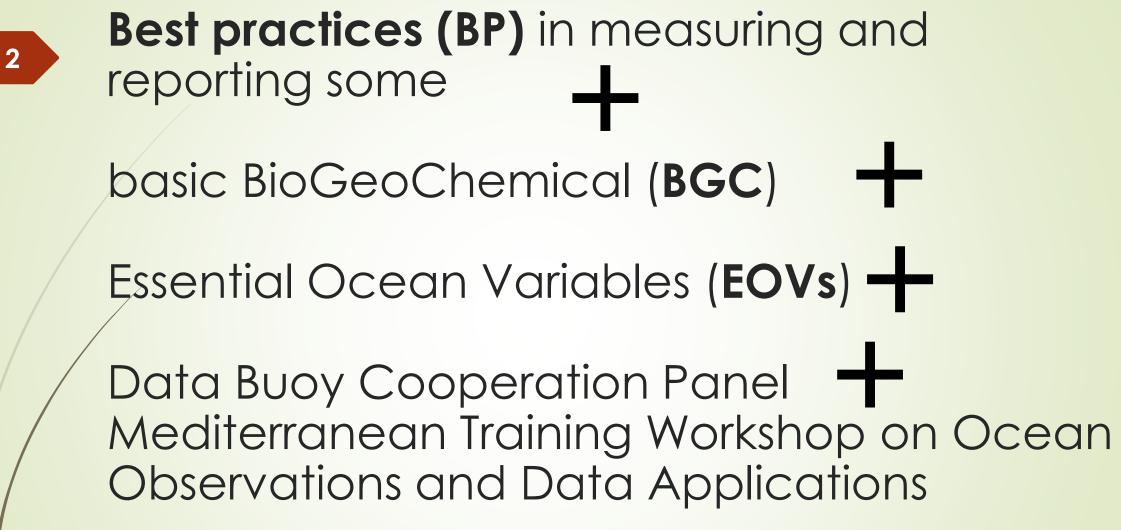
# Best practices in measuring and reporting some basic Biogeochemical Essential Ocean Variables

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# Introduction

3





# Introduction

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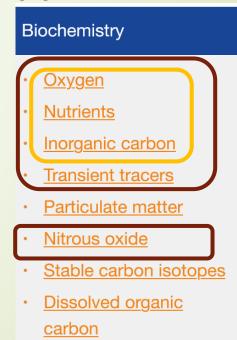
basic BGC C:N:P 17.4 hytoplankto \* Diazotrophs Aicrozoopian T Mesozoopiani 010 Faecal pellets 272

# • Perennial

Best practices: commercial or professional procedures that are accepted or prescribed

- <u>Impact</u>: the variable addresses climate, operational ocean services and ocean health and also contributes to understanding from a scientific perspective or application from a service perspective.
- <u>Feasibility</u>: it is technically, politically and economically feasible to observe or derive the variable on a global scale using proven, scientifically understood methods.

• Some also are **Essential CLIMATE** variables: characterize Earth Climate



as being correct or most effective





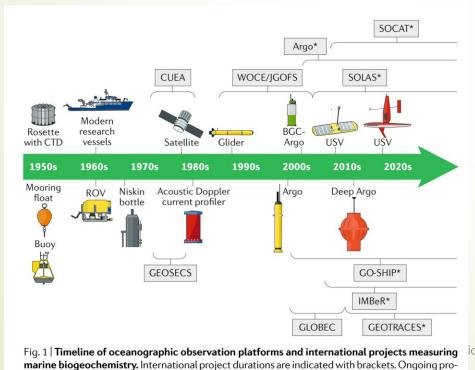
First DBCP Mediterranean Training Workshop on Ocean Observations and Data Applications-Part 2, Tunis, 2-4 May 2023

GOOS web page

# Introduction

#### **Basic BGC EOVs**

- Historically measured
- Sound BP
- Involved in all ocean processes
- Useful at any temporal / spatial scale
- Sensor development, to be installed in different platforms
- Discrete measurements always needed



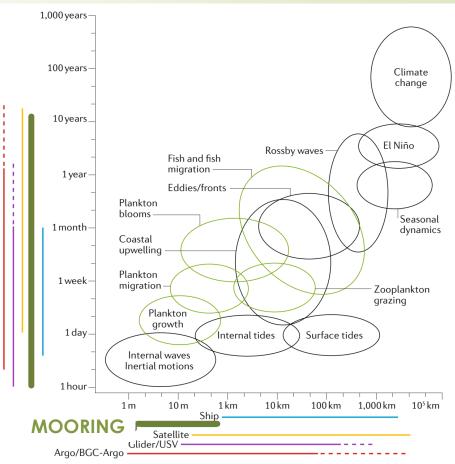


Fig. 2 | **Measuring across spatiotemporal scales in marine systems.** Spatial and temporal scales of marine dynamic and ecosystem processes, and the measurement capabilities of different observational platforms. Dynamic and/or physical processes are represented by black circles and biological and/or ecological processes by green circles. The presence of new autonomous platforms, such as

Chai et al 2020

ications-Part 2, Tunis, 2-4 May 2023

n

Biochemistry

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<u>Oxygen</u>

<u>Nutrients</u>

Inorganic carbor

- Transient tracers
- Particulate matter
- <u>Nitrous oxide</u>

ceanographic

<u>Stable carbon isotopes</u>

WMO

Dissolved organic
 <u>carbon</u>

# OUTLINE

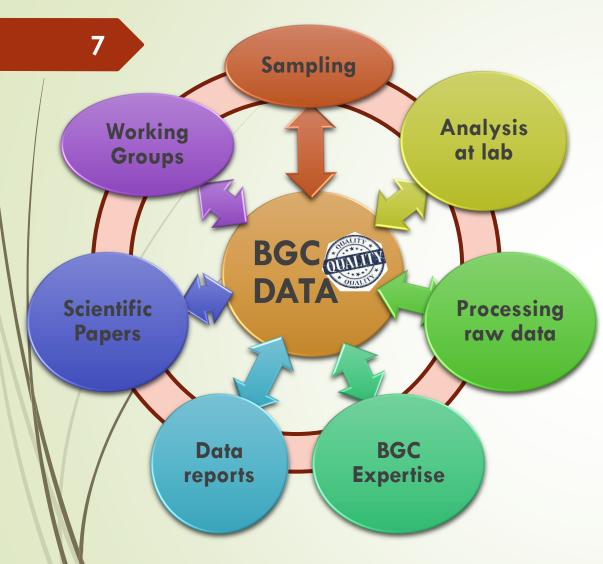


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- **1. Common best practices for bBGC EOVs**
- 2. Specific BP for dissolved oxygen
- **3.** Specific BP for inorganic nutrients
- 4. Specific BP for CO2 variables
- **5. Easy reporting, formatting: metadata & data**
- 6. Easy primary Quality Control (QC)
- 7. Examples of BAD practices / results



# **1. Common best practices for bBGC EOVs**



<u></u>

unesco

### SAMPLING & STORAGE & TRANSPORT

- As important as having reference materials!!
- Read the Standard Operational Procedure (SOP)
- Adapt the SOP to your circumstances
- Check all reagents prepared first
- Sampling = Training, training and training
- Make a video and never trust a newcomer
- Prepare your SOP, your <u>WIKI</u> with all the details (hard work but useful!!)
- Prepare reproducibility exercises (competition) for the samplers
- At the beggining qualified technitians need to KEEP an EYE on the sampling procedures around the rosette for example
- Water is terrible it escapes! Flows through the tiniest holes...
- Transport companies can be mad; FOAM!, PARAFILM, bubble wrap



REF: Langdon (2010)

#### SAMPLING

Flared neck flasks with long stopper – should be volume calibrated referred to 20°C

GO-SHIP Repeat Hydrograph

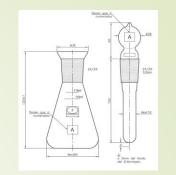
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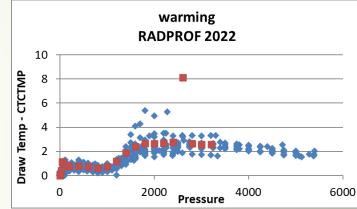
Reports and Guidelines. IOCCP Report No. 14 ICPO Publication Series No. 13

 Note the draw temperature – important for deep stations TMP differences > 2.5°C ≈ 0.1% in O2

### **STORAGE & TRANSPORT**

- Cool and dark place, water in the flared neck
- Cumbersome transport!, no turn around the BOX
- Maximum storage 1-2 months, but not recommended







unesco



### **REF: Langdon (2010)**

#### ANALYSING

- Air conditioning if possible, record temperature in the lab
- Be aware of CHEMICAL RISKS
- Bring the sampling sheets
- Use a notebook to mark and take NOTES
- Check the **bubbles** in the line of the titration solution (run junk samples at beginning)
- Prepare your standard with KiO3 or buy it from OSIL
- Blanks are a pain but needed
- All yolumetric material needs to be calibrated at least every two years.
- Make long long batch (same day many samples) of analysis to profit the standarization



- No internal laboratory standard or substandard can really be used!
- NOTES on every change of reagents







#### PROCESSING

- Excel or whatever but the same checked template
- Check the notes
- Report on international units for dissolved oxygen CONTENT ... umol /kg .. MASS units

### **REPORTING (for your shelf!)**

The report can also be a template with detailed information .. But always the same information

### **REPORTING (for others!!)**

- Minimum and concise information

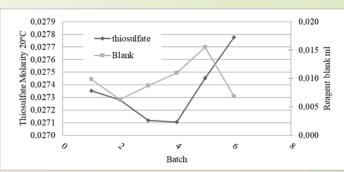
#### (1) OXYGEN

Potentiometric titration with Metrohm Titrando system with 5 ml burete coupled with a Metrohm Dosino, blanks determined, following Langdon (2010),

standarization with OSIL KIO3 0.01 N, precision better than  $\pm 1\ \mu\text{mol/kg}$  from replicate analysis

 Table 1. Temporal evolution of the thiosulfate concentration and reagent blank along the cruise. CV stands for Coefficient of Variation and STD for standard deviation.

Thiosulfate	Thiosulfate standardizations		Reagent Blank		Ν
batch	Molarity20°C	CV*100	ml	STD*100	analises
0 1	0.027354	0.12	0.010	0.48	81
1	0.027282	0.09	0.006	0.05	50
2	0.027119	0.01	0.009	0.28	87
3	0.027107	0.15	0.011	0.26	97
3	0.027454	0.18	0.016	0.28	78
o 4	0.027776	0.15	0.007	0.06	84
	batch           0         1           0         1           0         2           0         3           0         3	Iniosunate batch         Iniosunate Molarity20°C           0         1         0.027354           0         1         0.027282           0         2         0.027119           0         3         0.027107           0         3         0.027454	Infosurate batch         Infosurate Molarity20°C         CV*100           0         1         0.027354         0.12           0         1         0.027282         0.09           0         2         0.027119         0.01           0         3         0.027107         0.15           0         3         0.027454         0.18	Iniosunate batch         Iniosunate Molarity20°C         CV*100         ml           0         1         0.027354         0.12         0.010           0         1         0.027282         0.09         0.006           0         2         0.027119         0.01         0.009           0         3         0.027107         0.15         0.011           0         3         0.027454         0.18         0.016	Infostitute         Infostitute         Infostitute         Infostitute         Infostitute           0         1         0.027354         0.12         0.010         0.48           0         1         0.027282         0.09         0.006         0.05           0         2         0.027119         0.01         0.009         0.28           0         3         0.027107         0.15         0.011         0.26           0         3         0.027454         0.18         0.016         0.28



**Table 2**. Dissolved oxygen determinations (umol.kg<sup>-1</sup>) for replicated samples at stations 18, 76,108 and 109.

Station_cast_btl	Press	Salinity	O2 umol/kg	Diff O2	Diff Sal
18_16_5	1399.96	35.4957	219.11		
18_16_6	1400.50	35.4948	219.64	0.53	0.0009
18_16_7	1400.23	35.4946	219.24	0.13	0.0011
18_16_8	1400.60	35.4946	219.18	0.06	0.0000
18_16_9	1400.32	35.4952	219.12	0.07	0.0006
109_11_2	4749.52	34.8955	243.26		
109_11_3	4749.95	34.8955	242.78	0.48	0.0000
76_24_6	300.575	35.5720	213.02		
76_24_6	300.575	35.5720	212.96	0.06	0.0000
76_24_6	300.575	35.5720	210.68	2.28	0.0000
76_24_6	300.575	35.5720	214.05	3.37	0.0000
76_24_6	300.575	35.5720	209.71	4.34	0.0000
108_12_2	4399.62	34.8972	242.41		
108_12_3	4399.76	34.8972	242.36	0.05	0.0000



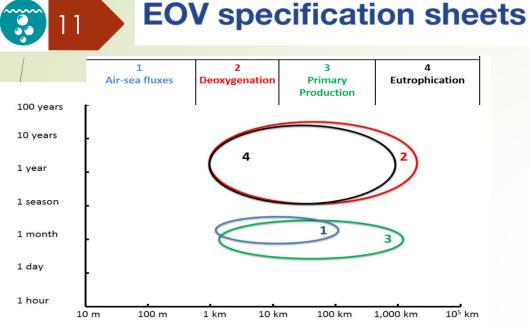
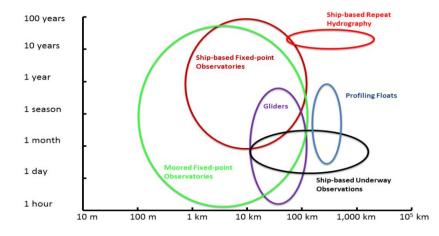
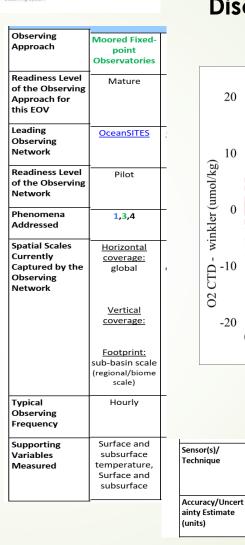


Figure 1: Spatial and temporal scales of phenomena (as color-coded and listed in Table 2 above) to be addressed.

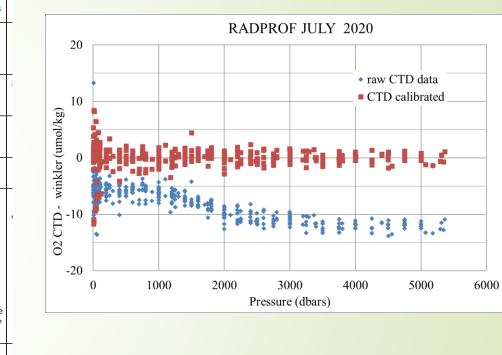




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1 200

#### Discrete meas. always needed!



Optical oxygen

sensor

±2.0

µmol O<sub>2</sub> kg<sup>-1</sup>



Figure 2. Spatial and temporal observation scales of component networks listed in Table 3 (thick coloured circles) and in Table 4 (thin black circles).

Unesco Intergovernmental Oceanographic

# **3. Specific BP for inorganic nutrients (uM, NO<sub>3+2</sub>, PO<sub>4</sub>, SiO<sub>2</sub>)**



REF: Becker et al. (2020)



- New sterile plastic High Density Polyethylene (HDPE) or Polypropylene (PP) (12-20 mL)

GO-SHIP Repeat Hydrograph

fanual: A Collection of Experimental Experts and Guidelines.

- Recycling tubes / containers is possible .. Cleaning investment
- NO smokers allowed
- Better NO gloves, that any gloves, .... Best .. Powder free vinyl gloves
- Avoid spigot from niskin bottles, or use clean silicon tube
- Buy a labeler is worthy!!

### **STORAGE & TRANSPORT**

- Particles interferences: centrifugation or filtering (0.45 um filter)
- Analysis of FRESH samples is highly recommended .. But sometimes impossible
- Best option up to now: freezing ...
  - / Leave a headspace for expansion of seawater
  - Tight caps
  - Always upright till really frozen
  - Buy many tube racks (sampling, organizing, freezing) Ziploc bags to store sets of tubes in horizontal Other options of storage to be tested in <u>EuroGO-SHIP</u>









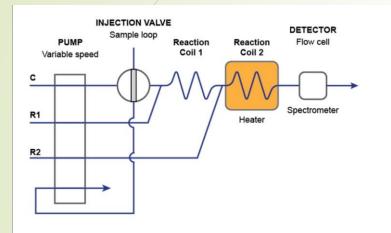






#### REF: <u>Becker et al. (2020)</u> and also

Highly reccomended to understand PRINCIPLES of the Continous Flow Analysis for inorganic nutrients in seawater => Aminot and Kerouel (2007)



### ANALYSIS

Oceanographic

- N2 carrier / pumps / glassware / detector
- Many reagents
- Primary / secondary / working (mixed) standards
- Waste management







Kah Arent, Roje Ketted Dosage automatique des nutriments dans les eaux marines







#### **QUALITY CONTROL AND ASSESSMENT**

Quality control—The overall system of activities whose purpose is to control the quality of a measurement so that it meets the needs of users. The aim is to ensure that data generated are of known accuracy to some stated, quantitative degree of probability, and thus provides quality that is satisfactory, dependable, and economic.

Quality assessment—The overall system of activities whose purpose is to provide assurance that quality control is being done – effectively. It provides a continuing evaluation of the quality of the analyses and of the performance of the analytical system.

#### Internal control using

- Tracking sample / substandard lab prepared
- External , bought as CRMs or RMs (KANSO, KIOST, MOOS) optimal to our content range !!
- MedSea .. Very low uM so difficult!!



#### Final control using

- Primary QC visual inspection, prop-prop plots
- Secondary QC comparison to other data in área / time

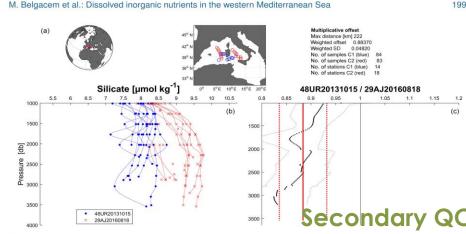
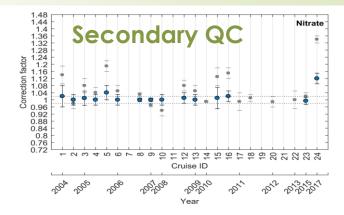


Figure 4. An example of the calculated offset for silicate between cruise 48UR20131015 and cruise 29AJ2016818 (reference cruise). (a) Location of the stations that are part of the crossover and statistics. (b) Vertical profiles of silicate data ( $\mu$ mol kg<sup>-1</sup>) of the two cruises that fall within the minimum distance criteria (the crossing region), below 1000 dbar. (c) Vertical plot of the difference between both cruises (thick dotted black line) with standard deviations (dbited red lines).



**Figure 5.** Results of the crossover analysis for nitrate, before (grey) and after (blue) adjustment. Error bars indicate the standard deviation of the absolute weighted offset. The dashed lines indicate the 2% accuracy limit for an adjustment to be recommended.





#### **Cruise Reports**

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The following should be included in the nutrient section of the cruise reports:

- (i) Cruise designation (ID) and principle investigator(s).
- (ii) If not listed in the cruise report elsewhere, CTD station information including station position, time, sampling depths, bottle numbers etc.
- (iii) Names and affiliations of the analysts.
- (iv) Numbers of samples analyzed, batches of standards used, pump tube and column changes.
- (v) Equipment, methodology, and reagents used.
- (vi) Sampling and any storage procedures.
- (vii) Calibration standard information, methods, and values.
- (viii) Data collection and processing procedures.
- (ix) Details of any problems and trouble-shooting that occurred.
- (x) QC/QA:
  - stated accuracy and analytical precision;
  - detection limits;
  - values of check samples and/or tracking standards;
  - measured values of the reference materials (including which batch was used, and assigned or certified values);
  - if and how normalizations were made to the data, based on the internal check/tracking samples or the CRM.

#### (xi) Scientific References.

#### Becker et al. (2020)





#### **REFERENCE MATERIALS : LOW AND HIGH CONCENTRATION**

**Table 1.** Mean and standard deviation (STD) values for the #CE and #CG RMN analysis performed for the 5 RADPROF batch of analysis.

RADPROF RMN # CE	Mean±STD μmol/kg	certified value µmol/kg
Nitrate	0.02±0.03	0.01±0.03 <sup>*</sup>
Nitrite	0.022±0.01	0.016±0.009
Silicate	0.00±0.32	0.06±0.09 <sup>*</sup>
Phosphate	0.030±0.036	0.012±0.006 <sup>*</sup>

RADPROF RMN # CG	Mean±STD μmol/kg	certified va µmol/kg
Nitrate	24.4±0.34	23.7±0.2
Nitrite	0.06±0.01	0.06±0.03
Silicate	57.431±0.7	56.4±0.5
Phosphate	1.67±0.07	1.70±0.02

#### INTERNAL LAB SUBSTANDARD: coastal water filtered

Mean and standard deviation (STD) values for the surface seawater analysed for the 5 RADPROF run of analysis.

RADPROF SW	Mean±STD μmol/kg	
Nitrate	10.84±0.11	
Nitrite	0.03±0.03	
Silicate	7.55±0.20	
Phosphate	0.33±0.02	

#### 5 runs => 4 weeks

 $\sim$ 16 

> Moored Fixed-Point Observatories

Resolving water column

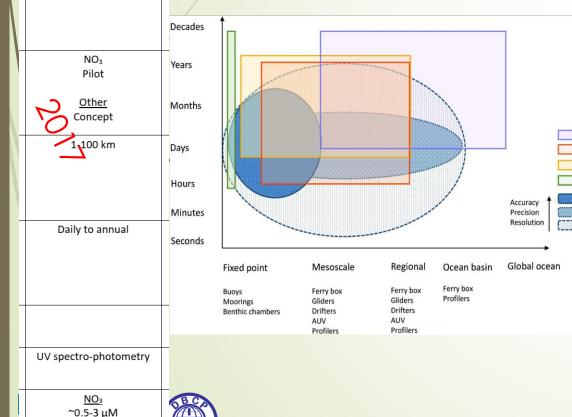
measurements of

nutrients.

# **EOV** specification sheets



#### Toward a Harmonization for Using in situ Nutrient Sensors in the Marine Environment Daniel et al. (2020)



		Advantages	Disadvantages	Optimal conditions of deployment
		Wet chemical analyzers (spectroph	otometry and fluorimetry)	
		Limit of detection Low trueness Low precision: 1 – 4% Large concentration range for each nutrient In situ calibration Moderate response time (few min.) Not significantly affected by biofouling	High power requirements: 1, 5 – 28 W Relatively bulky: 500 × 200 mm, 5 – 8 kg High maintenance costs Longevity limited by reagent use and stability (1 – 5 months) Hazardous waste chemical storage	Coastal and open ocean, relatively short time periods (days to months) Temperate
		UV optical sensors		
ors	Measurement frequency	Good trueness Large nitrate concentration range Reagent free Fast response: few seconds – 2 min Low power requirement: 4.2 – 8 W Moderate weight: 1.8 – 3.4 kg	Nitrate only Biofouling Subject to optical interferences Nitrate only Moderate limit of detection Moderate trueness	Coastal waters and open ocean, long time period (months to year), polar temperate and tropical
		Electrochemichal sensors		
		Moderate silicate concentration range Reagent free Moderate weight: 2.2 kg Not significantly affected by biofouling	Moderate limit of detection Moderate trueness	Coastal waters and open ocean, relatively short time period (days to months), polar

All nutrients

Moderate response time

Not significantly affected by biofouling Moderate precision: 2% Subject to large variation of salinity Long measurement cycle (30 - 60 min) Benchmark techniques in laboratories

Only discrete samples High trueness and precision Potential contamination during the Not affected by biofouling sampling Substantial and consistent maintenance Not optimized for long-term unattended operation

#### Wish List

Optimal conditions of Tested deplo

platforms

FerryBox

Benthic chami

Fixed platform

BLIOV

Gliders

Moorings

Buoy

Moorings

Buoys

Profiling floats

open ocean, long time Profiling floats

period (months to year), Gliders

to months), polar

temperate and tropical

- APNA, ChemFIN (Subchem systems, C Egli et al., 2009)
- WIZ (SYSTEA, Moscetta et al., 2009)
- Micromac C (SYSTEA)
- Hydrocycle PO4 (Seabird scientific)
- NAS3X (Mills et al., 2005) - Digiscan (Plant et al., 2009)
- ANAIS (Thouron et al., 2003)

P

P

P

C

- Alchemist (Le Bris et al., 2000)
- CHEMINI (Vuillemin et al., 2009a)
- LoC (Beaton, 2012; Nightingale et al., 2015; Beaton et al., 2017; Clinton-Bailey et al., 2017; Grand et al., 2017; Vincent et al., 2018)
- NuLAB (GreenEyes)
- ISUS/SUNA (Seabird scientific; Johnson and Coletti, 2002; Johnson et al., 2006; D'Ortenzio et al., 2012, 2014; Frank et al., 2014)
- ProPS/OPUS (TriOS; Zielinski et al., 2011; Meyer et al., 2018)
- Spectro:lyser (S::can Measuring Systems, Etheridge et al., 2014) - SUV-6 (Finch et al., 1998)
- ANESIS (Lacombe et al., 2008; Jońca P et al., 2013; Barus et al., 2016, 2018)
- AA3 and QuAAtro SEAL analytical - SkalarVR. - Lachat QuikChem

\* The authors of the references cited did not necessarily use the same way to calculate the Limit Of Detection. P, prototype; 0 First DBCP Mediterranean Training Workshop on Ocean Observations and Data Applications of the reference and add not necessary use the same way to cal

Drifters and profilers

Ship of opportunity

Coastal time series

Electrochemical sense

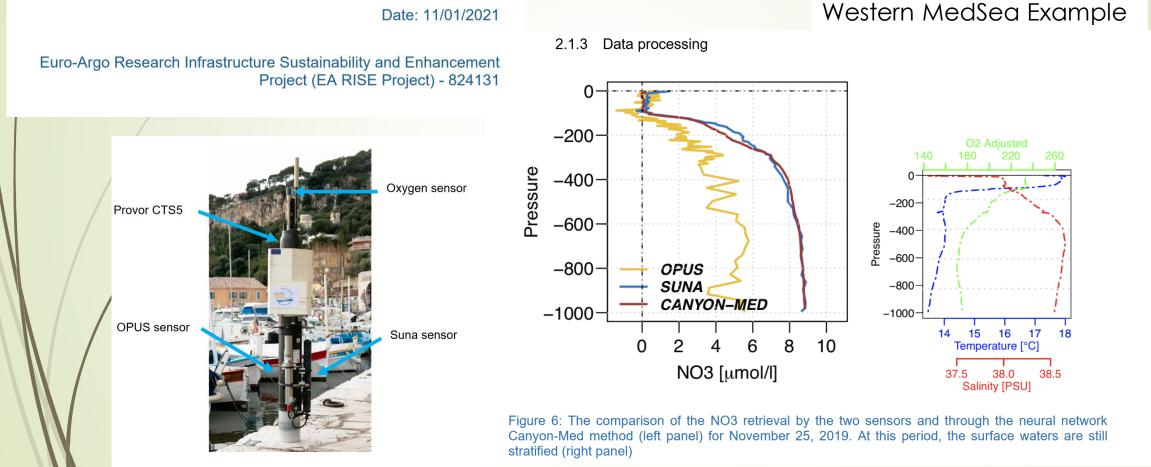
Wet chemistry

Optical sensors

Gliders



#### Discrete meas. always needed? .. Probably yes!





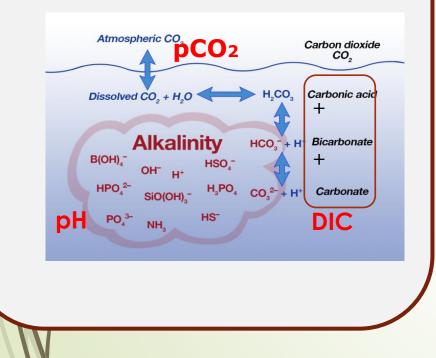
Oceanographic Commission



unesco

#### **Basic concepts!**

Seawater CO<sub>2</sub> system => <u>acid base system</u> of minor constituent in seawater (<0.7%, <200 mg)



### REF: Dickson et al (2007)

Table 8. Estimates of the Analytical Precision and Accuracy of Measurements of pH, TA, TCO<sub>2</sub>, and pCO<sub>2</sub>

analysis	precision	accuracy	ref
pH (spectrophometric)	$\pm 0.0004$	$\pm 0.002$	42
TA (potentiometric)	$\pm 1 \mu { m mol}{ m kg}^{-1}$	$\pm 3 \mu  m mol~kg^{-1}$	29
TCO <sub>2</sub> (coulometric)	$\pm 1 \mu$ mol kg <sup>-1</sup>	$\pm 2 \mu \text{mol kg}^{-1}$	96
$f_{\rm CO_2}$ (infrared)	$\pm 0.5 \mu \text{atm}$	$\pm 2 \mu \text{atm}$	97



#### CO<sub>2</sub> software for calculated variables:

 Table 1. Carbonate system software packages.

Package	Language	Version	Reference
CO2SYS <sup>a</sup>	QBasic	1.05	Lewis and Wallace (1998)
CO2SYS <sup>b</sup>	Excel	24	Pelletier et al. (2007)
CO2SYS <sup>a</sup>	Excel	2.1	Pierrot et al. (2006)
CO2SYS <sup>a</sup>	MATLAB	1.1	van Heuven et al. (2011)
CO2calc <sup>c</sup>	Visual Basic	1.3.0	Robbins et al. (2010)
csys <sup>d</sup>	MATLAB	04-2014	Zeebe and Wolf-Gladrow (2001)
ODV <sup>e</sup>	C++	4.5.0	Schlitzer (2002)
mocsy <sup>f</sup>	Fortran 95	2.0	Orr and Epitalon (2015)
seacarb <sup>g</sup>	R	3.0.6	Gattuso et al. (2015)
swco2 <sup>h</sup>	Excel	2	Hunter (2007); Mosley et al. (2010)
swco2 <sup>h</sup>	Visual Basic	2	Hunter (2007)

<u>Orr et al. 2015</u>



Guide to Best Practices for

Ocean CO<sub>2</sub> Measurements



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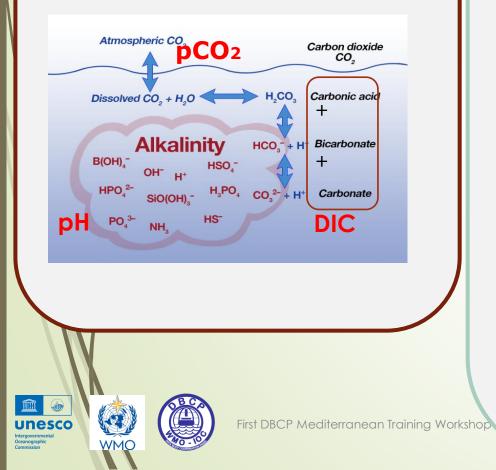
Riebesell et al. (2011)

First DBCP Mediterranean Training Workshop on

19

#### **Basic concepts!**

Seawater CO<sub>2</sub> system => <u>acid base system</u> of minor constituent in seawater (<0.7%, <200 mg)



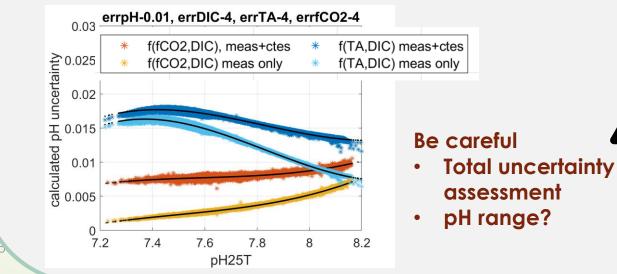
#### CO<sub>2</sub> software for calculated variables:

Table 9. Estimated Probable Errors in the CalculatedParameters of the Carbonate System Using Various InputMeasurements

	input	pН	$TA \\ (\mu mol \ kg^{-1})$	$TCO_2$ ( $\mu$ mol kg <sup>-1</sup> )	f <sub>co2</sub> (µatm)
-	pH-TA			±3.8	±2.1
	pH-TCO <sub>2</sub>		$\pm 2.7$		$\pm 1.8$
	pH-fco,		$\pm 21$	$\pm 18$	
	$f_{\rm CO_2}$ -TCO <sub>2</sub>	$\pm 0.0025$	±3.4		
	$f_{\rm CO_2}$ -TA	$\pm 0.0026$		$\pm 3.2$	
l	TA-TCO <sub>2</sub>	$\pm 0.0062$			±5.7

<u>Orr et al. 2018</u>

#### Calculated pH





unesco

WMO

ceanographi

## All ocean processes affect CO2 variables

QSR OSPAR, <u>McGovern et al (2023)</u>

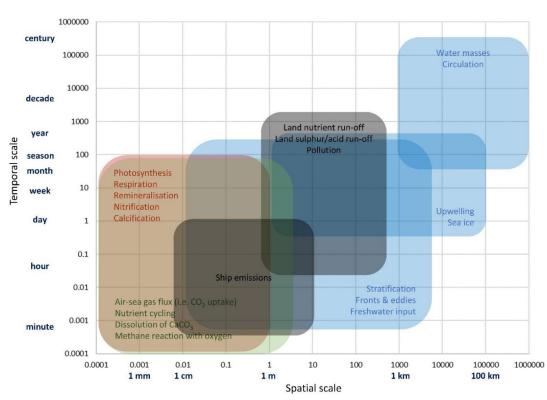
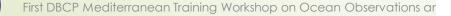
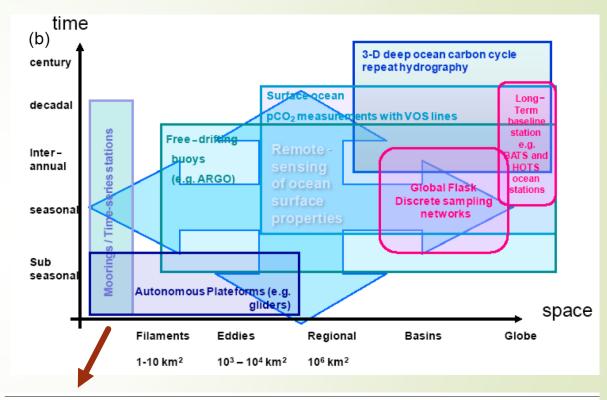


Figure 3.3: Simplified schematic overview on physical (blue boxes), chemical (green boxes), biological (red boxes) and anthropogenic (black boxes) processes that contribute to changes in ocean carbonate chemistry on different temporal and spatial scales.



#### Observing requirements, Ciais et al. (2014)



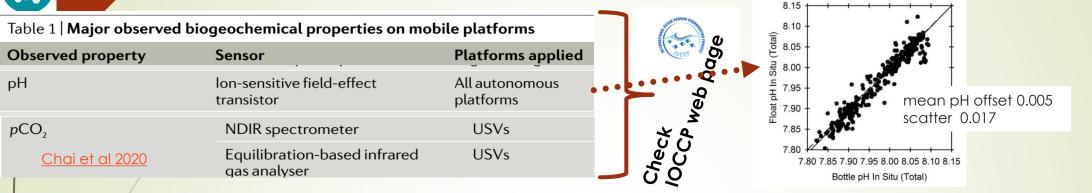
Chai et al 2020

Observed property	Sensor	<b>Platforms</b> applied
pCO <sub>2</sub>	NDIR spectrometer	USVs
	Equilibration-based infrared gas analyser	USVs
рН	lon-sensitive field-effect transistor	All autonomous platforms

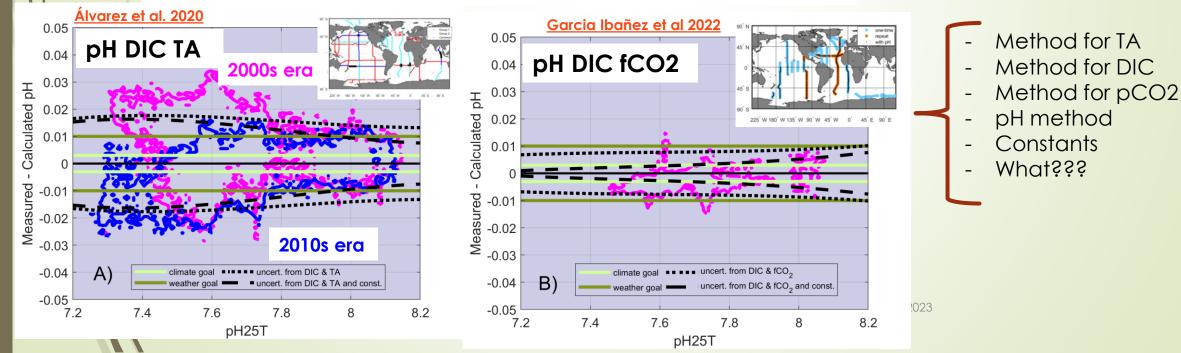


#### Observing requirements, Ciais et al. (2014)

Very promising tech (Johnson et al. 2017) BUT



#### NEED to go back to DISCRETE MEASUREMENTS and CHECK





### pH or not pH ... ?;???

#### 1990s

- Introduction of sulfonephtalein indicator dyes to measure pH
- First characterization of m-cresol purple (Kodak mCP) and description of manual method
- WOCE era, pH rarely measured in transoceanic sections
- WOCE A25 (1997) first spect pH measurements in IIM-CSIC group

<u></u>

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### 2000s

- 2007 SOP description
   CLIVAR era, more pH measurements
- First GLODAP (2004) release, with no pH data included
- Development of automated systems for benchtop and submersible autonomous systems
- Detection of impurities in mCP dyes

#### 2010s

- Introduction and first characterization of purified mCP (1)
- Other characterizations and evaluation of uncertainties in pH method
- Use of ISFET electrodes in seawater
- GO-SHIP era, pH is EOV
  2009 109 first cruise
- 2009 109 first cruise with PUR mCP

#### 2020s

- GLODAP.v2 2016 release with pH data –mostly UNPUR mCP data, in 2021 both
- Measured and calculated pH disagreement => GLODAP applies corrections to reach consistency (2)
- Wider salinity and temperatura range PUR mCP characterization
- A434\_imp <u>Method to correct</u> UNPUR <u>to</u> PUR <u>mCP</u> (no paper published using it) (3)
- Lab tests to check agreement between ISFET and PUR pH
- Lab test to check stability and consistency between PUR mCP pH

#### Douglas & Byrne (2017)



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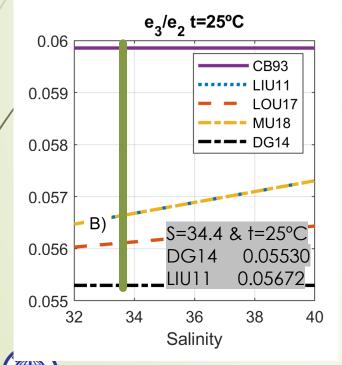
Álvarez et al. 2020 0.05 0.04 UNPUR pl 0.03 Hd Calculated 0.02 0.01 1 -0.01 Weasured -0.02 -0.03 PUR pH -0.04 A) INDEXESTIMATION OF A TAME IN TAME INTERNA NA A TAME INT uncert, from DIC & TA and const. -0.05 7.4 8.2 7.2 7.6 7.8 8 pH25T



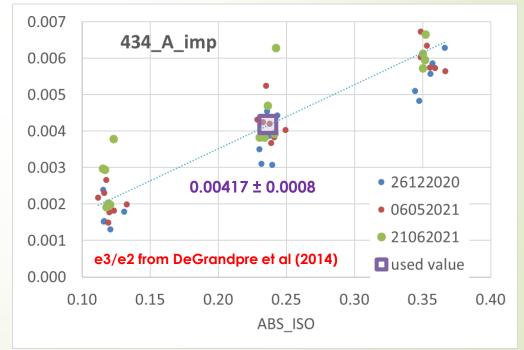
$${}^{434}_{imp}A = {}^{434}_{obs}A - \frac{e3}{e2} * {}^{578}_{obs}A$$

 $\simeq$  (0.06 Abs) - 0.05xx · ( $\simeq$  1 ABS)

<sub>434</sub> A <sub>imp</sub> => Nearly zero or negative !!!



#### UNPUR Sigma Aldrich #11517KC





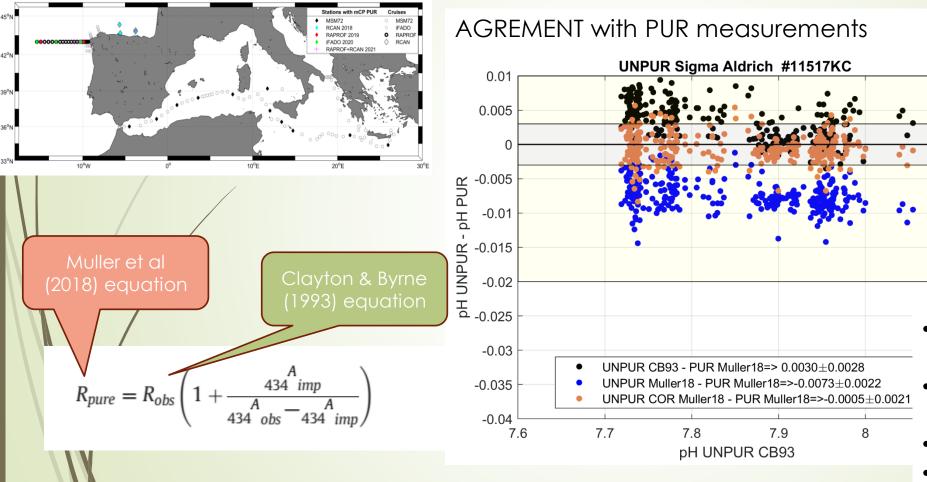
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WMC

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UNPUR Sigma Aldrich #11517KC  $^{434}_{imp}A = 0.00417 \pm 0.0008$ 

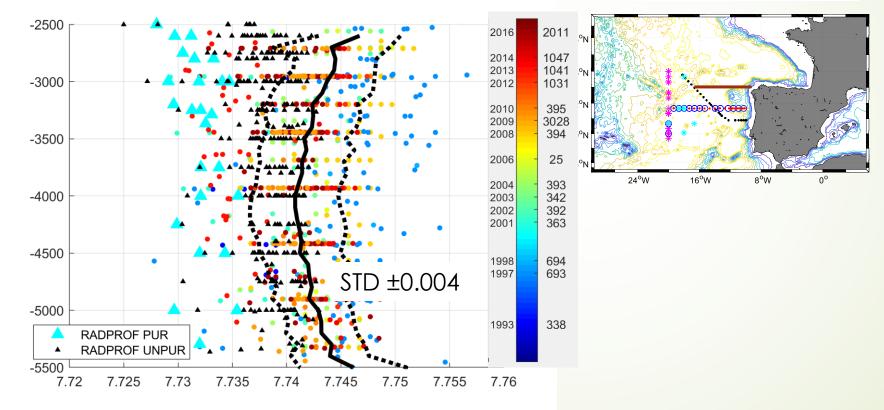




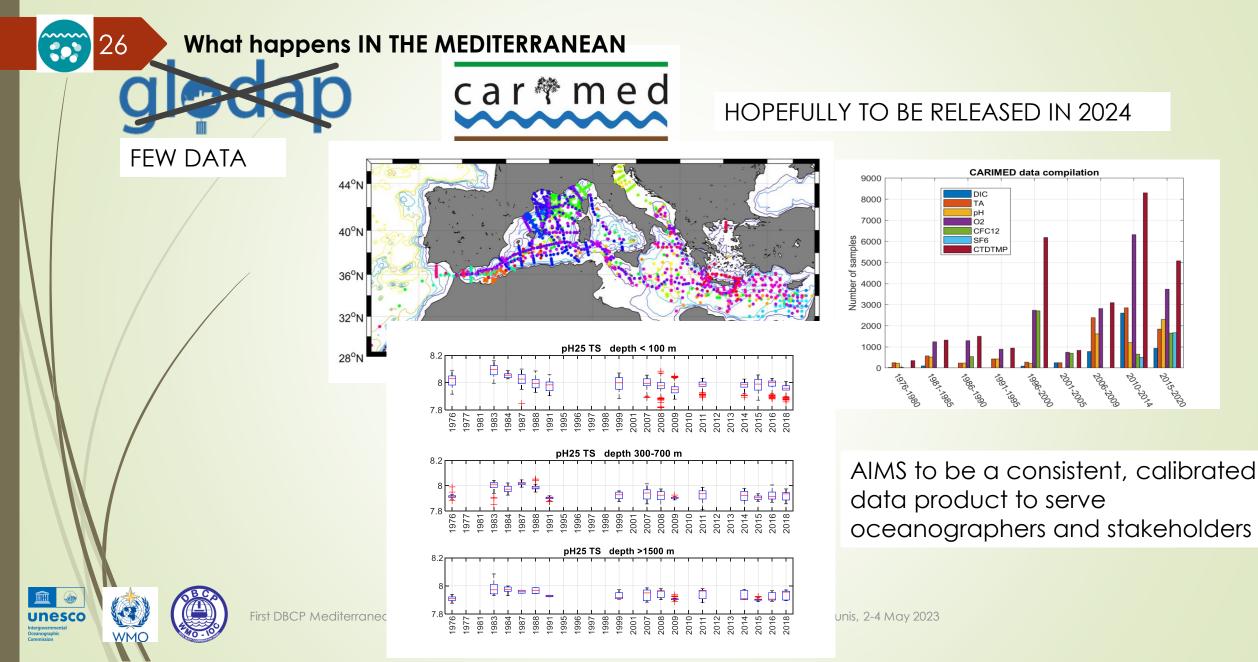
- Need to (even) check purified mCP
- Formerly used UNPUR mCP is needed to obtain <sub>434</sub>A<sub>imp</sub>
- Lab work
- Computer work to recover UNPUR mCP ABS and DeltaR

What happens if no mCP is available??

NE Atlantic pH time series – Depth > 2500 meters







# 4. Easy reporting, formatting: metadata & data

# Some reccomendations

- Create your template in excel for data & metadata Data
  - Vocabulary, variables names
  - International units
  - Always the same format for basic time / space variables
  - Always the same significant units for physical and BGC variables
- Metadata

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- Basic info about cruise
- Basic info about each variable, methods, RMs?, precision and accuracy checks

### On line tool excel to CSV convertor by A. Velo (IIM-CSIC)

Read <u>Jiang et al. (2022)</u>



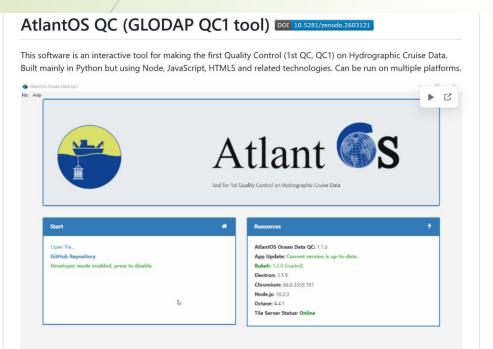
# 5. Easy primary Quality Control (QC)

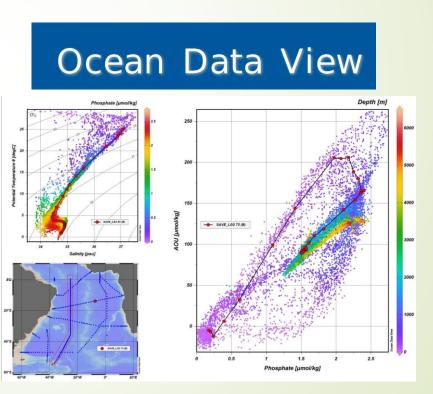
# Some reccomendations

Create your template in excel for checking data but it is tiresome Each data sample should be referenced Station / Cast / Niskin bottle

Tool by Velo et al reads created CSV file

<u>ODV</u> as well







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&

# **6. Examples of BAD practices / results**

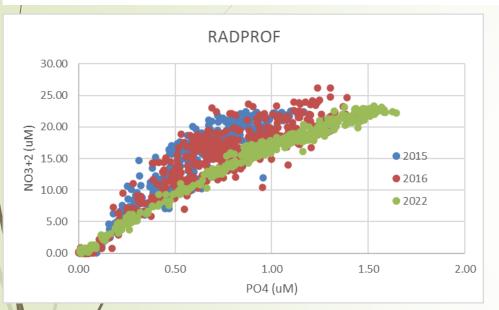
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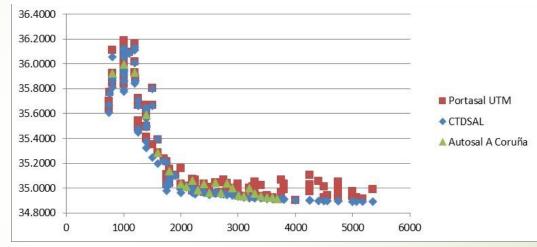
**WMO** 

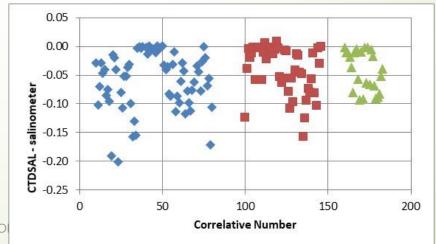
### INOCEN's fault!!

**NUT**: bad conservation .. Too long freezing



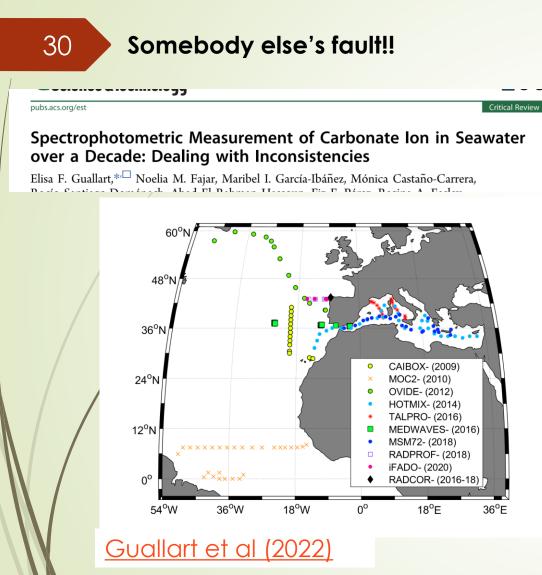
# **Salinity**: bad sampling/ conservation / analysis.. Who knows?



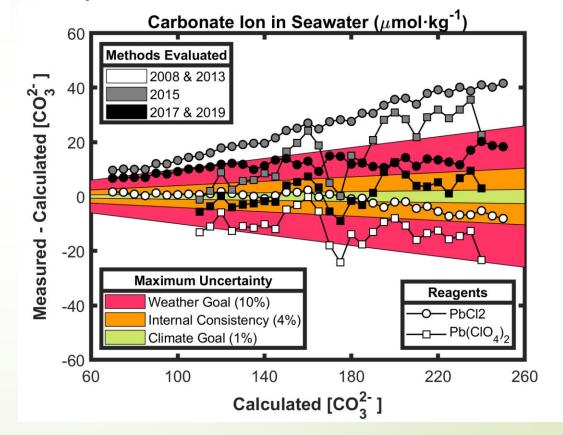


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# 6. Examples of BAD practices / results



#### Proposed BP not coherent with other CO2 variables

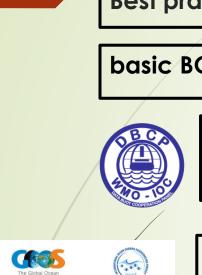




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## **7. CONCLUSIONS**

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Best practices: could be misleading

basic BGC but not easy

Plea for cheaper / easier to maintain / sustained BGC sensors

The Global Ocean



Hope for an easier access to CRMs

Hope for pH metrology

